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give rise to a number of problems in electrochemical experiments however a
basic understanding of the construction and operation of these electrodes
can prevent such problems and can improve their reliability'

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June 5th, 2020 - problem potential drop iR_u due to uncompensated solution
resistance actual interfacial potential is smaller than controlled
potential between we and ce $e_{int} = e_f - iR_u$ effect reduced current peak
flattened cv at apparently higher potential with dynamic iR_u correction
without correction quinhydrone 5 mm in 1 mm H_2SO_4 $v = 75$ mv s 1 jru' **'cyclic
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electrochemical measurement in a cyclic voltammetry experiment the working
electrode potential is ramped linearly versus time unlike in linear sweep
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