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**name Molecular 180 6 ×**

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**Objectives Section 10 1 Define**

**?mole? in terms of a quantity in**

**chemistry Know Avogadro?s**

**Number and how it relates to**

**the mole Understand the type of**

**representative particle for**

**elements molecular compounds**

**ionic compounds and ions see**

**Table 10 1 on p 290'**

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*Quantities 10 1 The mole a*

*measurement of matter*

*Avogadro?s number mass of a*

*mole 10 2 Mole mass and mole*

*volume relationships STP 10 3*

*Percent composition skip*

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**The Mole A Measurement of**

**Matter You often measure the**

**amount of something by count**

**by mass or by volume A mole**

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mol of a substance is  $6.02 \times 10^{23}$  representative particles of that substance'

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