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dipole moments are non polar b
Hydrogen?s electronegativity acts
similar to phosphorus because they
have'*

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Answers Both are polar and equally
symmetric but the difference in
electronegativity between N F is less
than that between P

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**Electronegativity and Polarity 265
Section 88 5 5 Objectives Describe
how electronegativity is used to
determine bond type Compare and
contrast polar'**

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**For our purposes use the
definitions in the chart The most
polar bond determines the polarity
of a molecule i e if a compound
contains one non polar and one
polar bond the molecule as a whole
is considered to be polar'**

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